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Balancing Act Answer Key: Page 1 Problems 2 Ca + O 2 2 CaO N 2 + 3 H 2 2 NH 3 2 Cu 2O + C 4 Cu + CO 2 2 H 2O 2 2 H 2O + O 2 Hint: Add the O atoms on the product side together when doing the counts. Page 2 Practice Problems 1. 2 Na + MgF 2 2 NaF + Mg 2. Mg + 2 HCl MgCl 2 + H 2 3. Cl 2 + 2 KI 2 KCl + I 2 4.

Name Atoms are not or during a chemical reaction. Mg + O

Mg of the chemical formulas in the equation. You cannot 1) Determine number of atoms for each element. 2) Pick an element that is not equal on both sides of the equation. 3) Add a coefficient in front of the formula with that element and adjust your counts. 4) Continue adding coefficients to get the same number of atoms of each element Mg on each side.

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Balancing Equations: Answers to Practice Problems 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers).

Balancing Equations: Practice Problems

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Balancing Act Worksheet 1 Answers

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